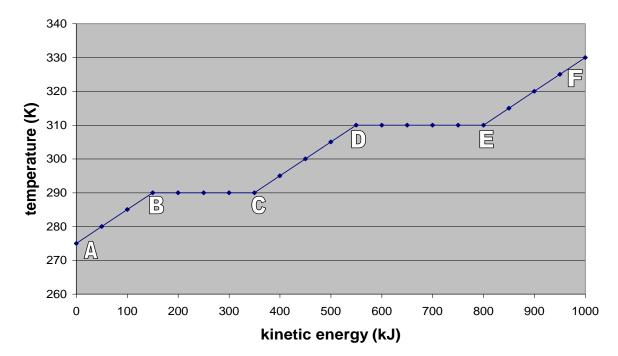
Use the following graph to answer the questions that follow.



Phase Changes for a Substance: Temperature vs. Kinetic Energy

- 1. What phase is this substance in at point A?
- 2. What phase is this substance in at point F?
- 3. What is the general term for what is happening at points $B \rightarrow C$ and points $D \rightarrow E$?
- 4. What specifically is happening at points $B \rightarrow C$?
- 5. What specifically is happening at points $D \rightarrow E$?
- 6. What specifically is happening at points $C \rightarrow B$?
- 7. What specifically is happening at points $E \rightarrow D$?
- 8. At what point does this substance begin to freeze?
- 9. At what point does this substance begin to melt?
- 10. At what point does this substance begin to boil?
- 11. At what point does this substance begin to condense?
- 12. What is the freezing point of this substance?
- 13. What is the melting point of this substance?
- 14. What is the boiling point of this substance?
- 15. Describe an actual physical situation this graph may be representing (disregard exact temperatures).